

# Algorithmic Techniques for Atomistic Modeling

Hasan Metin Aktulga  
Dept of Computer Science  
Purdue University  
haktulga@cs.purdue.edu

May 14, 2009

**ayg: chage green to a darker color**

# Outline

1. **Introduction: MD Methods & Reactive Force Fields**
2. **Our Implementation of Reactive Force Fields (ReaxFF)**
  - General structure of a ReaxFF implementation
  - Implementation details of major components
  - Additional features
3. **Applications using the Serial Version**
  - Hexane simulations: Validation and performance analysis
  - Water-Silica Surface Interaction
  - Si/Ge Nanobar Strain Rates
4. **Ongoing & Future Work**
5. **Parallelization of ReaxFF**
  - Challenges
  - Our solution approaches
6. **Current Userbase**

# Introduction: Atomistic Modeling Methods

- **Ab-initio methods:** Atomistics with electronic degrees of freedom
  - Hartree-Fock(HF) methods → misses  $E_{corr}$
  - Post-HF methods
    - \*  $E_{corr}$  incorporated at great computational expense
    - \* semi-empirical methods to the rescue
  - DFT-based methods
    - \* completely different approach but similar derivations to HF theory
    - \* bigger systems, longer simulation times made possible
    - \* CPMD is the most popular example
- **Classical MD methods:**
  - many approximations: no electronic d.o.f.
  - electronic effects are mimicked through parametrizations
  - static bonds, no reactions!
  - systems at nanoscale, simulation times upto hundreds of ns
- **Continuum Mechanics:** Macroscale systems
  - additional approximations, no atomistic detail!
  - modeled using PDE's.

↑  
accuracy  
complexity  
detail  
comp.  
demand

approx's  
sys size  
sim.  
frame  
↓

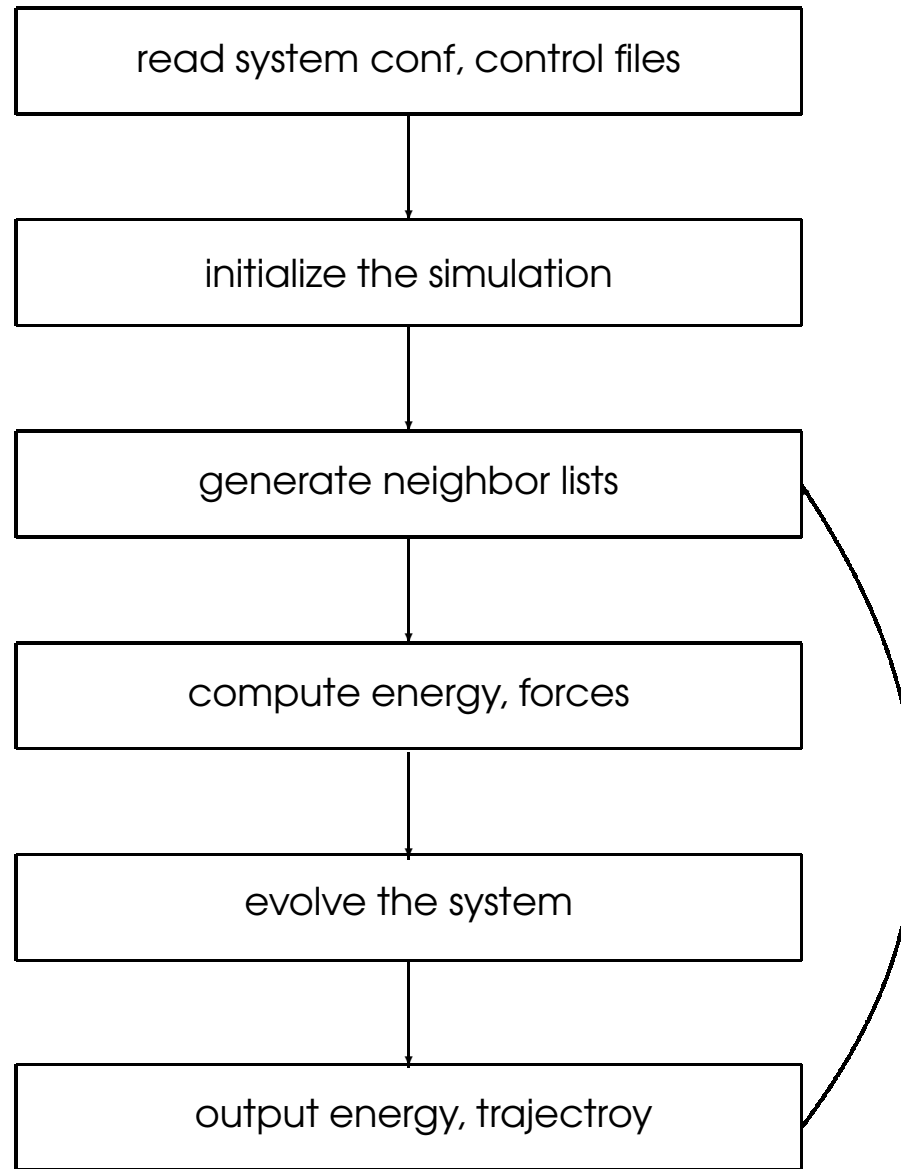
## Reactive Force Fields (ReaxFF): Bridging the Gap

	ReaxFF	Classical MD
<b>advantages</b>	reactive	non-reactive
	dynamic bonds	static bonds
	polarization with QEq	fixed charges in general (except for polarizable FF)
<b>challenges</b>	dynamic interaction lists	static lists
	complex & costly enegies,forces	much simpler energy & force formulas
	frequent update of charges (expensive!)	static charges
	shorter timesteps ( $\approx 0.25$ fs)	longer timesteps (1 – 10 fs)

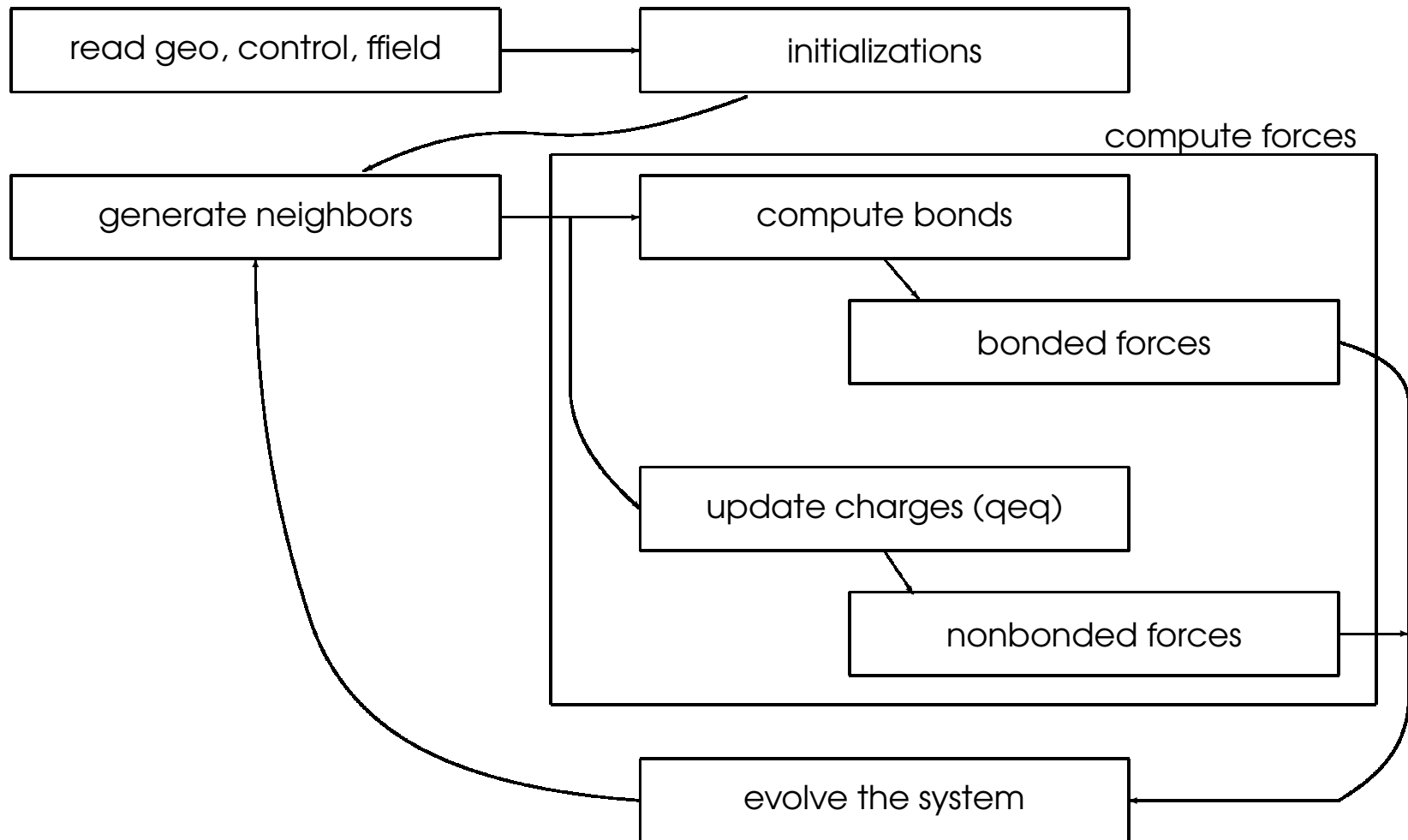
Interesting applications: large system with reactions and charge-transfer

Simulation of fuel cells, silica crack propogation, corrosion of silica in water, etc.

## General Flowchart of Conventional MD Programs



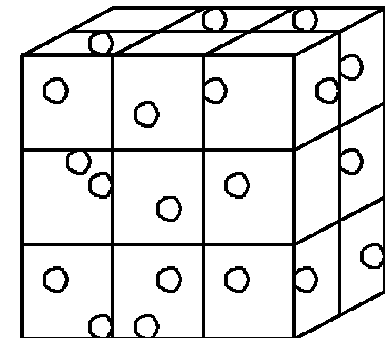
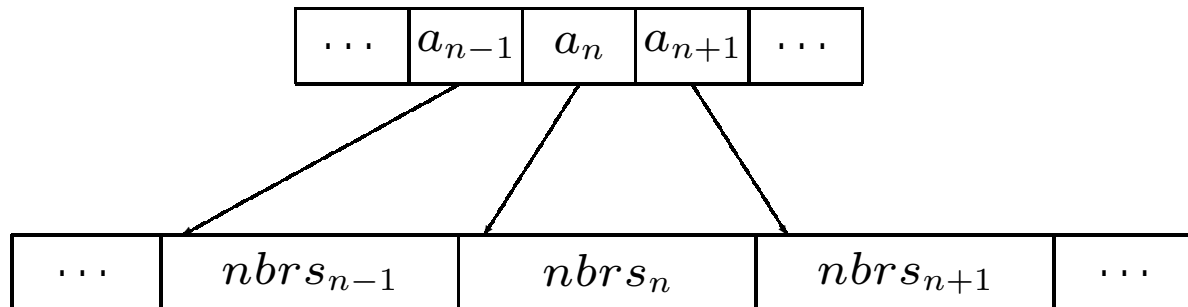
## ReaxFF Flowchart



ayg: What about valence corrections?

## Implementation: Neighbor Generation

- 3 different neighbor lists:
  - `near_nbrs` for bonded forces → `bond_cut`  $\approx 4\text{-}5\text{ \AA}$ , full matrix stored
  - `hbond_list` for hydrogen bonds → `hbond_cut`  $\approx 6\text{-}7.5\text{ \AA}$ , only for  $H$
  - `far_nbrs` for non-bonded forces → `nonb_cut`  $\approx 10\text{ \AA}$ , upper-half only
- Bin atoms into 3D grid cells
  - grid cell dims  $\approx \frac{1}{2}\text{nonb\_cut}$
- Verlet lists with delayed re-neighboring **not** implemented → little benefit
- Compressed adjacency list representation



# Implementation: Computing Forces and Potentials

- **Bonded Interactions:** Similar to classical MD but accounts for dynamic bonds
  - precursor: bond orders
  - lone-pair energy, over/undercoordination energies
  - bond energy
  - valence energy (with penalty & 3-body conjugation corrections)
  - dihedral energy (with 4-body conjugation correction)
- **Hydrogen Bonds**
  - precursor: bond orders
  - $H$  covalently bonded to  $X$  and interacting with  $Z$
  - can be considered a bonded interaction
- **Non-bonded Interactions**
  - precursor: charge equilibration (QEq)
  - electrostatic (Coulomb) energy, van der Waals energy

$$\begin{aligned} E_{system} = & E_{bond} + E_{lp} + E_{over} + E_{under} + E_{val} + E_{pen} + E_{3conj} \\ & + E_{tors} + E_{4conj} + E_{H-bond} + E_{vdW} + E_{Coulomb} \end{aligned}$$



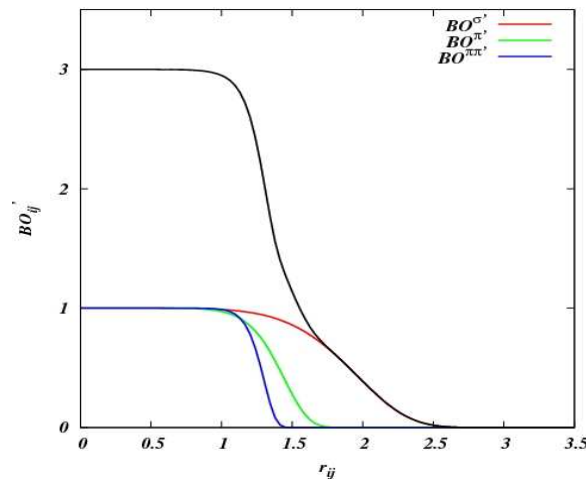
## Bonded Interaction: Bond Orders

- Prior to **bonded forces**, compute **bond orders** based on the new **near\_nbrs**
  - **bond\_list**: subset of **near\_nbrs**, stored in the same way
  - **uncorrected** bond orders and derivatives
  - store both  $\text{bo}(i, j)$  and  $\text{bo}(j, i) \rightarrow$  efficient construction of **angles**, **dihedrals**
  - compute  $\text{bo}(i, j)$  only if  $i < j$ , otherwise  $\text{bo}(i, j) = \text{bo}(j, i)$

$$\text{BO}'_{ij} = \text{BO}^{\sigma'}_{ij} + \text{BO}^{\pi'}_{ij} + \text{BO}^{\pi\pi'}_{ij}$$

$$\text{BO}^{\alpha'}_{ij}(r_{ij}) = \exp \left[ a_{\alpha} \left( \frac{r_{ij}}{r_{0\alpha}} \right)^{b_{\alpha}} \right]$$

$$\text{BO}_{ij} = \text{BO}'_{ij} \cdot f_1(\Delta'_i, \Delta'_j) \cdot f_4(\Delta'_i, \text{BO}'_{ij}) \cdot f_5(\Delta'_j, \text{BO}'_{ij}) \text{ where } \Delta'_i \text{ is the valency of atom } i.$$



## Bonded Interaction: Bond Energy

$$E_{\text{bond}} = -D_e^\sigma \cdot \text{BO}_{ij}^\sigma \cdot \exp \left\{ p_{be1} \left( 1 - \left( \text{BO}_{ij}^\sigma \right)^{p_{be2}} \right) \right\} \\ - D_e^\pi \cdot \text{BO}_{ij}^\pi - D_e^{\pi\pi} \cdot \text{BO}_{ij}^{\pi\pi}$$

- the **stronger** the bond, the **lower** the associated energy
- sweep over the **bond\_list**
- compute **bond\_energy** between  $i, j$  only if  $i < j$

# Bonded Interaction: Lone-Pair & Over/UnderCoordination

- Lone-pair energy

- $\Delta_i^{lp} = n_{opt}^{lp} - n_i^{lp}$
- energy associated with **unpaired** electrons of an atom  $\rightarrow$  **zero** for a fully coordinated atom
- single-body interaction  $\rightarrow$  just sweep over **atom\_list**

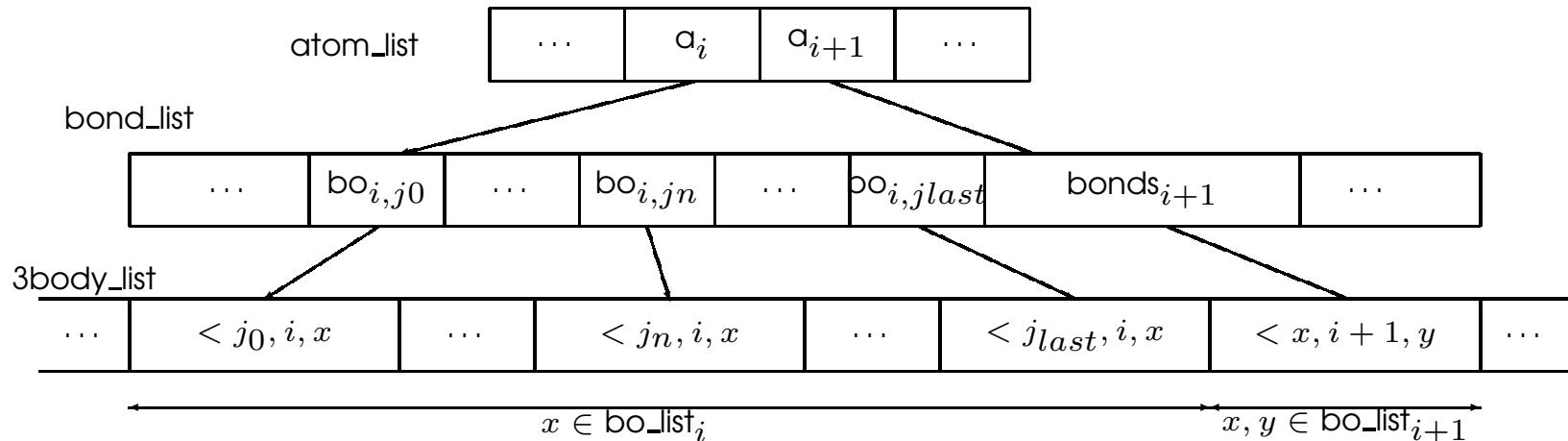
- Over/undercoordination energy

- **ideal** # of bonds = # of valence electrons
- $\Delta_i = \sum_{j \in nbrs(i)} bo(i, j) - Val_i$
- **actual** # of bonds  $>$  **ideal** # of bonds ( $\Delta_i > 0$ )  $\rightarrow$  over-coordination
- **actual** # of bonds  $<$  **ideal** # of bonds ( $\Delta_i < 0$ )  $\rightarrow$  under-coordination
- **actual** # of bonds = **ideal** # of bonds ( $\Delta_i = 0$ )  $\rightarrow$  **no** over/undercoordination energy
- functionals of  $\Delta_i$  and  $\Delta_j$ 's  $\rightarrow$  just sweep over **atom\_list**

## Bonded Interaction: Valence Angle Energy

$$E_{\text{val}} = f_7(\text{BO}_{ij}, p_{\text{val}3}, p_{\text{val}4}) \cdot f_7(\text{BO}_{jk}, p_{\text{val}3}, p_{\text{val}4}) \cdot f_8(\Delta_j, p_{\text{val}5}, p_{\text{val}6}, p_{\text{val}7}) \cdot \left( p_{\text{val}1} - p_{\text{val}1} \cdot \exp \left\{ -p_{\text{val}2} \cdot (\Theta_0 - \Theta_{ijk})^2 \right\} \right)$$

- $\Theta_0$  is the **ideal** angle,  $\Theta_{ijk}$  is the **actual** angle
  - the **closer** the  $\Theta_{ijk}$  to  $\Theta_0$ , the **lower** the energy
- $f_7(\text{BO}_{ij}, \dots)$  and  $f_7(\text{BO}_{jk}, \dots)$  ensure  $E_{\text{val}} \rightarrow 0$  as  $\text{BO}_{ij} \rightarrow 0$  or  $\text{BO}_{jk} \rightarrow 0$
- $\{\forall x, y \in \text{bo\_list}_i | x < y\}$  that meet certain criteria, compute the energy of  $< x, i, y$
- $\{\forall x, y \in \text{bo\_list}_i | x > y\}$  copy  $< y, i, x$  into  $< x, i, y$  (**for dihedrals!**)
- $E_{\text{pen}}$  and  $E_{3\text{conj}}$  account for corrections in special cases



## Bonded Interaction: Dihedral Energy

$$E_{tors} = \frac{1}{2} \cdot f_{10}(BO_{ij}, BO_{jk}, BO_{kl}, p_{tor2}, 1) \cdot \sin\Theta_{ijk} \cdot \sin\Theta_{jkl} \cdot V_{123}(\omega_{ijkl})$$

- Dihedral angle  $\omega_{ijkl}$  is the angle between planes defined by positions of  $i, j, k$  and  $j, k, l$
- $f_{10}(BO_{ij}, BO_{jk}, BO_{kl}, \dots)$  ensure  $E_{tors}$  vanishes smoothly as any of these bonds dissociate
- $\sin\Theta_{ijk}$  and  $\sin\Theta_{jkl}$  ensure that  $E_{tors} \rightarrow 0$  as  $\Theta_{ijk} \rightarrow 0$  or  $\Theta_{jkl} \rightarrow 0$
- $\{\forall i, j, k, l \in \text{atoms} | j < k, < i, j, k \in \text{3body\_list}_{jk}, < j, k, l \in \text{3body\_list}_{kj}\}$  compute the energy associated with  $\omega_{ijkl}$
- weak but very important in determining the 3D structures
- no higher order interactions  $\rightarrow$  no storage necessary

# Hydrogen Bonds

$$E_{XHZ} = p_{hb1} \cdot f_7(\text{BO}_{XH}, p_{hb2}, 1) \cdot \sin^4\left(\frac{\Theta_{XHZ}}{2}\right) \cdot \exp\left\{-p_{hb3} \cdot \left(\frac{r_{hb}^0}{r_{HZ}} + \frac{r_{HZ}}{r_{hb}^0} - 2\right)\right\}$$

- Constraints of a hydrogen bond:
  - Middle atom must be  $H$
  - $X, Z$  must be one of  $N, O, P, F$
  - $X - H$  covalently bonded,  $Z \in \text{hbond\_list}_H$
- $f_7(\text{BO}_{XH}, \dots)$  ensure  $E_{\text{val}} \rightarrow 0$  as the covalent bond breaks
- $\sin^4(\frac{\Theta_{XHZ}}{2})$  maximized when  $\Theta_{XHZ} = \pi$  ensures alignment on a line
- crucial for accurately describing water, DNA structure, secondary structures in proteins, etc.

## Nonbonded Interaction: Charge Equilibration (QEq)

Minimizing electrostatic energy by redistributing (partial) charges.

$$\text{Minimize } E(Q_1 \dots Q_N) = \sum_A (E_{A0} + \chi_A^0 Q_A + \frac{1}{2} J_{AA}^0 Q_A^2) + \sum_{A < B} (J_{AB} Q_A Q_B)$$

$$\text{subject to } Q_{net} = \sum_{i=1}^N Q_i$$

- Solve the optimization problem using the method of Lagrange multipliers
  - gives a sparse linear system of equations for finding charges
- GMRES with restarts,  $GMRES(50)$ 
  - heavy diagonal  $\rightarrow$  diagonal preconditioner
  - little configurational change between steps  $\rightarrow$  initial guess  $q_t = \text{linear\_extrapolation}(q_{t-1}, q_{t-2})$
- Implemented GMRES both with *MGS* and *Householder* orthogonalizations
  - virtually no difference
- Implemented CG for comparison
  - GMRES takes fewer number of matvecs and is faster than CG
- Choose tolerance for the norm of the relative residual carefully:
  - too high  $\rightarrow$  wrong results!
  - too low  $\rightarrow$  QEq dominates the total computation time!
  - more on this later ...

## Nonbonded Interactions: Coulomb & van der Waals Energy

$$E_{Coulomb} = C \cdot Tap(r_{ij}) \cdot \frac{q_i \cdot q_j}{[r_{ij}^3 + \gamma_{ij}^{-3}]^{\frac{1}{3}}}$$

$$E_{vdWaal} = Tap(r_{ij}) \cdot D_{ij} \cdot \left[ \exp \left\{ \alpha_{ij} \cdot \left( 1 - \frac{f_{13}(r_{ij})}{r_{vdW}} \right) \right\} - 2 \cdot \exp \left\{ \frac{1}{2} \cdot \alpha_{ij} \cdot \left( 1 - \frac{f_{13}(r_{ij})}{r_{vdW}} \right) \right\} \right]$$

- **Shielding** prevents energies from increasing drastically at close distances
- Long range interactions with **cutoffs**
  - **Taper** term ensures smooth vanishing of energies after the cutoff
- **no 1 – 2, 1 – 3 or 1 – 4 exclusions** → smooth bond forming/breaking
- Takes up a **large portion** of the total computation time
  - **tabulate** long range energy & forces
  - **linear interpolation**
  - large table → good approximations, reasonable memory usage
  - more on our gains later ...



## Summing Alltogether: Net Force

- Let  $E_i$  be the sum of energies from all interactions involving atom  $i$
- Let  $r_i$  denote the position of atom  $i$
- $F_i = \frac{\partial E_i}{\partial r_i}$
- **Problem:**
  - bonded energy expressions include  $\text{BO}_{ij}(\text{BO}'_{ij}, \Delta'_i, \Delta'_j)$  terms
  - $\frac{\partial \text{BO}_{ij}}{\partial r_k}$  arise in every bonded interaction
  - $\frac{\partial \text{BO}_{ij}}{\partial r_k} = c_1 \cdot \frac{\partial \text{BO}'_{ij}}{\partial r_k} + c_2 \cdot \frac{\partial \Delta'_i}{\partial r_k} + c_3 \cdot \frac{\partial \Delta'_j}{\partial r_k}$
  - $\frac{\partial \text{BO}_{ij}}{\partial r_k} \neq 0, \forall k \in \text{bonds}_i \cup \text{bonds}_j$ , **huge memory overhead!**
  - even if we choose to store them, **very time consuming** to compute each single bonded interaction!

## Summing Alltogether: Net Force - Solution

- **idea**: distribution law of multiplication over summation
- let  $C_0, \dots, C_n$  be the coefficients of  $\frac{\partial \text{BO}_{ij}}{\partial r_k}$  arising in different interactions
- re-write  $\sum_t C_t \times \frac{\partial \text{BO}_{ij}}{\partial r_k}$  as  $\frac{\partial \text{BO}_{ij}}{\partial r_k} \times \sum_t C_t$
- while computing interactions, accumulate  $C_t$ 's in  $C_{ijk}$
- **delay** computation of  $\frac{\partial \text{BO}_{ij}}{\partial r_k} \times \sum_t C_t$  and **forces** due to them until  $C_{ijk}$ 's are determined
- **no additional storage**, **important savings in CPU time**

## Implementation: Additional Features

- Modular implementation
  - a different force field can be adopted by plugging-in new interaction routines
- NVE, NVT and NPT ensembles **ayg: explain these terms first and what it takes to implement them**
- Compressible custom trajectory format
- Tools for performing common analysis
  - detection of reactions (on-the-fly)
  - property calculations such as drift coefficient, dipole moment (on-the-fly)
  - distributions of bond lengths, strengths, valence angles, charges, etc. (over the trajectory file)

# Applications

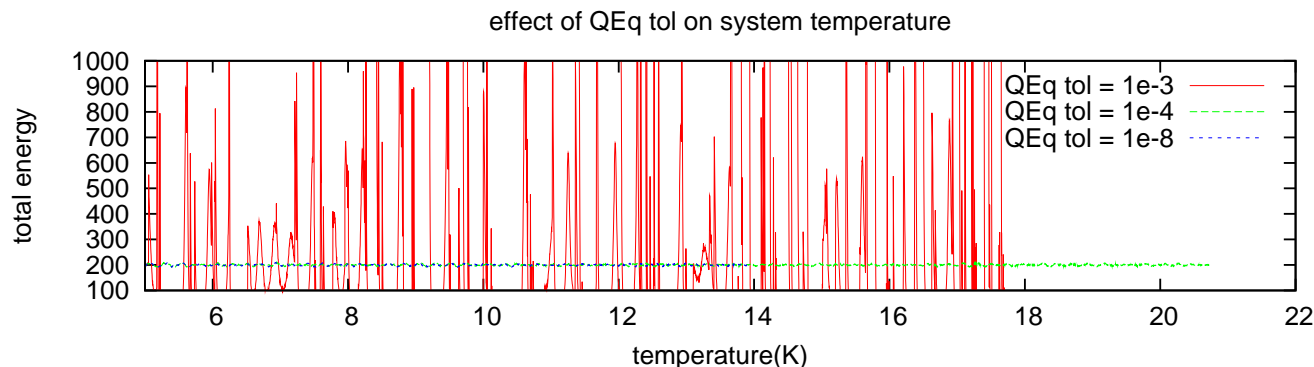
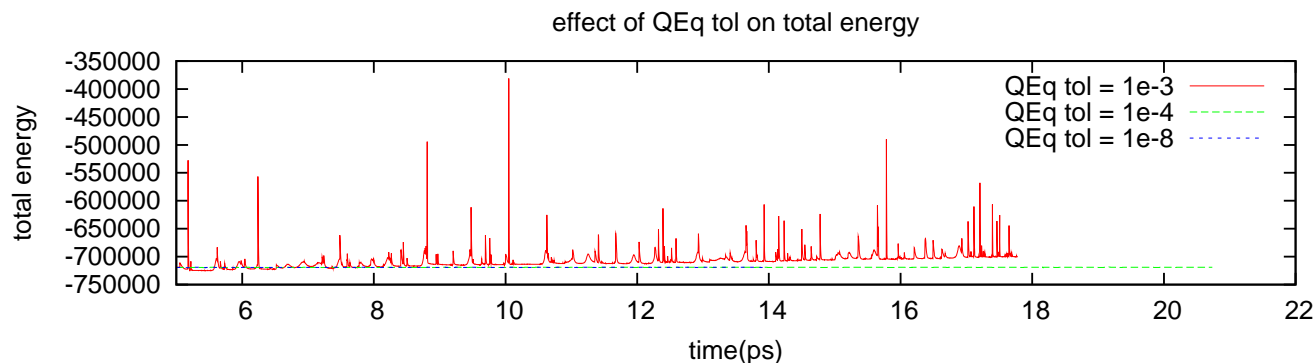
- Hexane Simulations: Validation and Performance Analysis
  - Preparation of systems
  - Effects of QEq tolerance on accuracy and performance
  - Effects of tabulating long range interactions on accuracy and performance
  - Hexane structure verification
  - Scalability of ReaxFF compared to *ab-initio* and classical MD
- Corrosion of silica surface in water (in collaboration with Dr Pandit's group)
- Measuring the strain tensor of *Si/Ge* nanobar (in collaboration with Dr Strachan's group)

# Hexane Simulations: Preparation of the Systems

- **Hexane:**  $C_6H_{14}$ , hydrocarbon, constituent of gasoline
- Initial configuration setup:
  - **Very large box** compared to the ideal volume → **reduces overlaps!**
  - **Randomly spread copies** of a model hexane molecule
  - **Rotations** of the model molecule around  $x, y$  and  $z$  axis to increase randomness
  - Various system sizes for scalability analysis (343, 512, 1000, 1728, 3375 molecules)
- Energy minimization and NPT simulations using **Gromacs**
  - brings the systems to the **ideal volume** quickly
  - **output** to be used for **ReaxFF** studies and scalability analysis
- Energy minimization and NVT equilibration using **ReaxFF**
  - added  $H$  to **Gromacs** output conf using the **Avogadro** program
  - energy minimization for 2.5 ps
  - NVT equilibration at 200 K for 2.5 ps
  - QEq tolerance set to  $1e-8$  to be safe

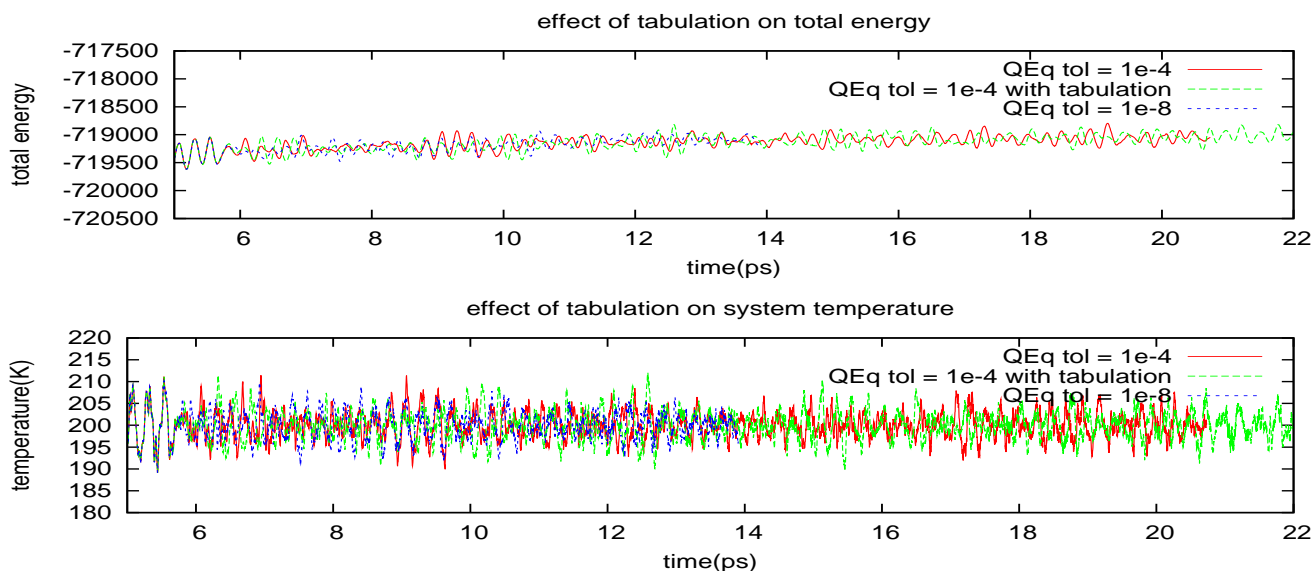
# Hexane Simulations: Effect of QEq Tolerance on Accuracy

- **Chosen:** hexane system with 343 molecules = 6860 atoms
- **Restart** from the system equilibrated at 200 K
- **How to determine the “right” tolerance:**
  - observe how the same system evolves over time at **different QEq tolerances**
  - pick **the highest one** with **reasonable accuracy**
  - $\text{tol}_1 = 1e-3, \text{tol}_2 = 1e-4, \text{tol}_3 = 1e-8 \rightarrow$  control run



# Hexane Simulations: Effect of Tabulation on Accuracy

$\text{tol}_2 = 1e-4$  looks good enough, now turn on tabulation of long range interactions, too!



More in depth comparison shows they are almost identical:

property	$\text{tol}_3 = 1e-8$	$\text{tol}_2 = 1e-4$	$\text{tol}_2 = 1e-4$ with opt.
C-H bond	$1.09 \pm 0.01$	$1.09 \pm 0.01$	$1.09 \pm 0.01$
C-C bond	$1.57 \pm 0.01$	$1.57 \pm 0.01$	$1.57 \pm 0.01$
<C-C-C	$108.0 \pm 2.9$	$107.9 \pm 2.9$	$108.0 \pm 2.9$
<C-C-H	$111.0 \pm 0.0$	$111.0 \pm 0.0$	$111.0 \pm 0.0$
<H-C-H	$106.6 \pm 0.0$	$106.6 \pm 0.0$	$106.6 \pm 0.0$
$q_{\text{C-tip}}$	-0.171	-0.171	-0.171
$q_{\text{C-mid}}$	-0.080	-0.080	-0.080
$q_{\text{H-tip}}$	0.040	0.040	0.040
$q_{\text{H-mid}}$	0.040	0.040	0.040

## Hexane Simulations: Profiling Analysis & Scalability

- Different qeq tolerances, with/without tabulation
- Used the *head.cs* cluster except for the first case

ayg: What are the numbers here.. are they times in seconds?

	total	neighbors	bonded	nonb	QEq	matvecs	QEq%
tol= $1e-4$ w/opt§	0.93	0.22	0.14	0.22	0.34	9.9	37%
tol= $1e-4$ w/opt	3.26	0.47	0.32	0.65	1.78	9.9	55%
tol= $1e-4$	4.41	0.45	0.31	1.92	1.67	9.9	38%
tol= $1e-6$ w/opt	3.35	0.45	0.31	0.59	1.97	13.6	59%
tol= $1e-6$	4.65	0.45	0.31	1.92	1.95	13.5	42%
tol= $1e-8$	7.56	0.44	0.30	1.91	4.89	46.8	65%

ayg: Dont say Dell Studio.. instead, say what the processor is, etc. §Architecture can make a huge difference. Same system with same parameters on a Dell Studio XPS with  $2.67GHz$  quad-core *i7* processor and  $1066MHz$  memory.

- Lessons learnt:
  - QEq tolerance is crucial for accuracy
  - arbitrarily large QEq tolerance might cause QEq domination ayg: What does the above bullet mean?
  - QEq is just a precursor to electrostatics yet takes up at least one third of total time!
  - QEq must be improved to make ReaxFF scalable.



## Hexane Simulations: Validation

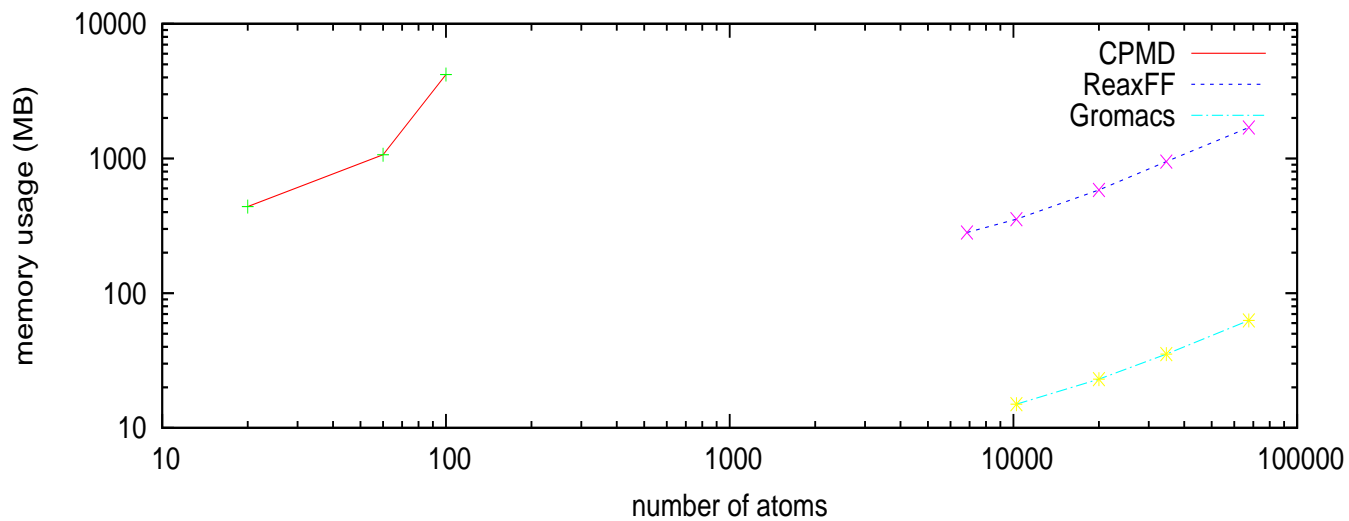
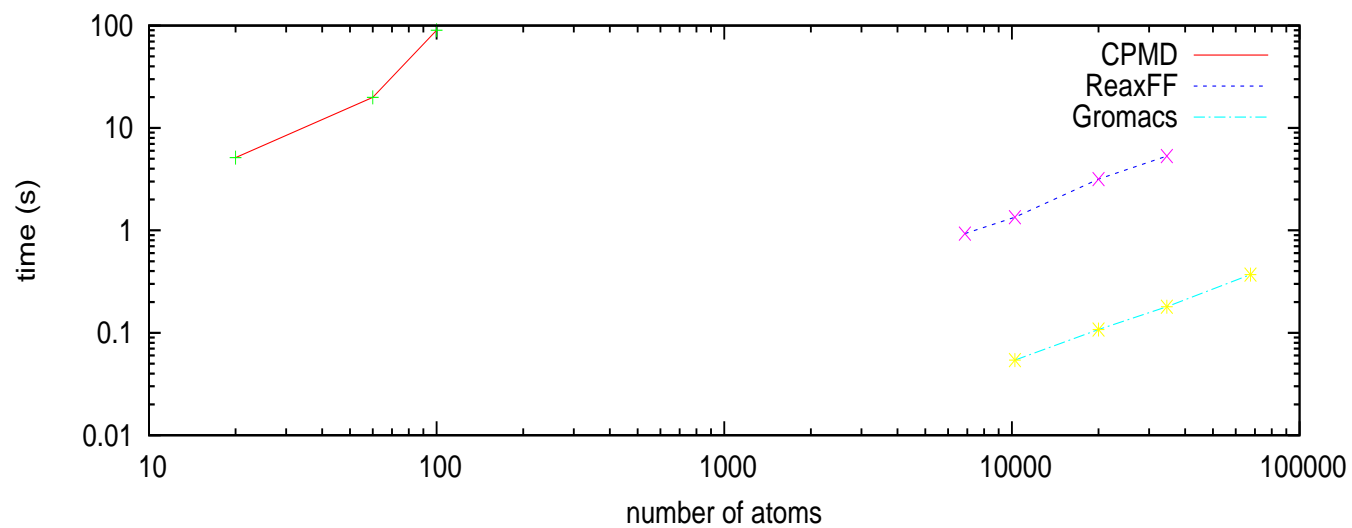
Compare the structure of our hexane molecules to those of [experimental results](#) in the literature and [ab-initio simulation](#):

property	ReaxFF	experimental <sup>†</sup>	<i>ab-initio</i> <sup>‡</sup>
C-H bond	$1.09 \pm 0.01$	$1.118 \pm 0.006$	1.100
C-C bond	$1.57 \pm 0.01$	$1.533 \pm 0.003$	1.533
<C-C-C	$108.0 \pm 2.9$	$111.9 \pm 0.4$	114.2
<C-C-H	$111.0 \pm 0.0$	$109.5 \pm 0.5$	109.5
<H-C-H	$106.6 \pm 0.0$	<i>NA</i>	106.5
$q_{\text{C-tip}}$	-0.171	<i>NA</i>	-0.205
$q_{\text{C-mid}}$	-0.080	<i>NA</i>	0.033
$q_{\text{H-tip}}$	0.040	<i>NA</i>	0.047
$q_{\text{H-mid}}$	0.040	<i>NA</i>	-0.10 ~ 0.10

<sup>†</sup>R. A. Bonham, L. S. Bartell, and D. A. Kohl. "The Molecular Structures of n-Pentane, n-Hexane and n-Heptane" J. Am. Chem. Soc., 1959, 81 (18), 4765 – 4769

<sup>‡</sup>geometry optimization of an isolated hexane using CPMD v3.13.2 with *PBE* Troullier-Martins pseudopotentials

# Hexane Simulations: Scalability Analysis



## Silica Surface Corrosion in Water

## *Si/Ge* Nanobar

## Ongoing & Future Work

- ParallelReax
  - complete & verify the implementation
  - a large scale application (maybe with the PRISM device)
- Integration into LAMMPS (a well-known, widely used MD package from SNL)
  - QEq integrated independently
    - \* opens the door for polarizable-ff inside LAMMPS
    - \* compatibility issues to be sorted out!
- Better solvers for QEq
  - block Jacobi type pre-conditioner
  - inner-outer schemes (use a lower cutoff inner solve to precondition outer solve)
  - use a fast multipole-type preconditioner.
- Tight relations among items on the agenda
  - better QEq solvers necessary for scalable ParallelReax
  - better QEq solvers necessary for QEq in LAMMPS
  - completion of ParallelReax necessary for Reax in LAMMPS

# Parallelization of ReaxFF

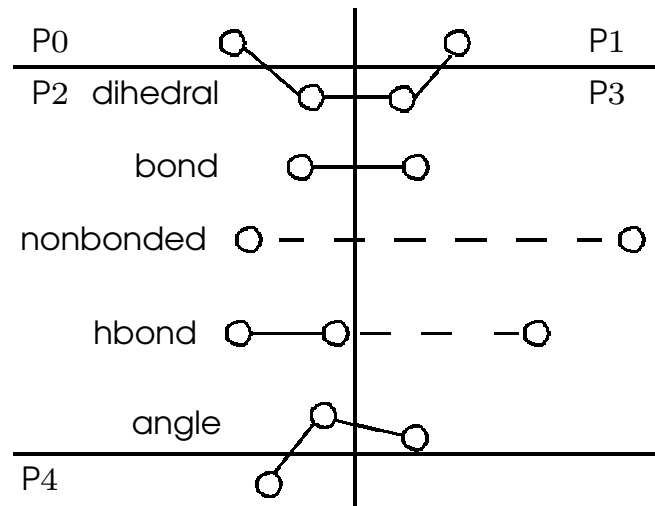
- A draft version of ParallelReax
  - domain decomposition technique
    - \* **repeat:** get my share of atoms → communicate boundaries → compute forces → move my atoms
  - not fully verified
  - inefficient handling of processor boundaries

## Two big challenges:

- Parallelization of QEq
  - even CG needs at least 4 communications per iteration!
  - parallel GMRES (with *Householder orth.*) is even worse
  - QEq will dominate even more
  - definitely need: better solvers for QEq
- Processor boundaries
  - avoid double computation at boundaries!
  - avoid thick boundaries, retain accuracy for any system!

# Parallelization: Our Solution Approaches

- **Parallelization of QEq:** better solvers for QEq
  - block Jacobi type pre-conditioner
  - inner-outer schemes
- **Avoid double computation**
  - coordination through **the mid-point rule**
    - \* **bond**( $i, j$ ): owner( $\frac{1}{2}(r_i + r_j)$ )
    - \* **<** ( $i, j, k$ ): owner( $r_j$ )
    - \* **dihedral**( $i, j, k, l$ ): owner( $\frac{1}{2}(r_j + r_k)$ )
    - \* **hbond**( $X, H, Z$ ): owner( $r_H$ )
    - \* **nonbonded**( $i, j$ ): owner( $\frac{1}{2}(r_i + r_j)$ )



# Parallelization: Our Solution Approaches

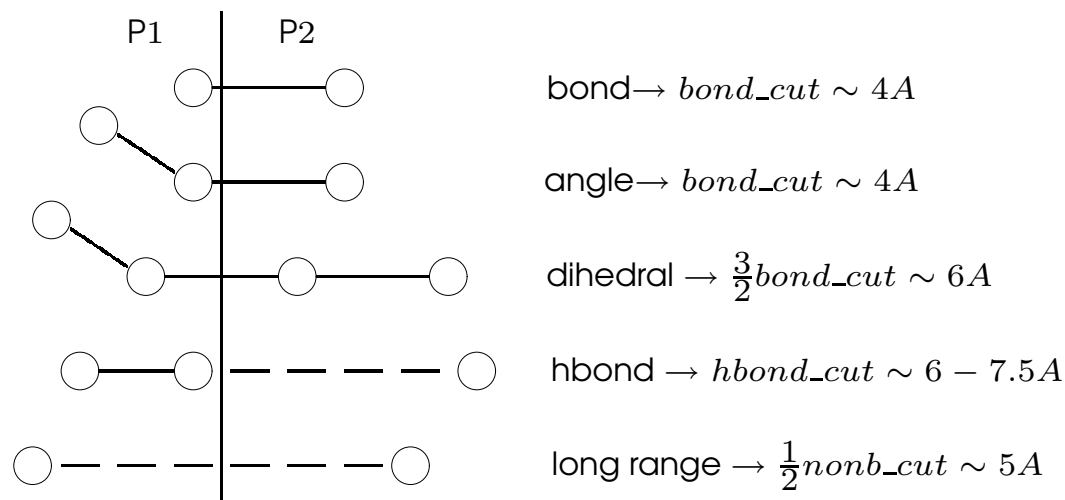
- Why avoid thick boundaries?

- a 20 Å cubic box  $\approx$  1000 atoms
- assume *nonb\_cut* long boundaries  $\rightarrow$  a few thousand atoms/processor  $\rightarrow$  upto a couple of seconds per iteration!
- definitely need: thin boundaries

- How to avoid thick boundaries?

- Mid-point rule to the rescue!
- boundary thickness:  $\max(\frac{3}{2}bond\_cut, hbond\_cut, \frac{1}{2}nonb\_cut)$
- gets better if no hydrogen bonds present:  $\max(\frac{3}{2}bond\_cut, \frac{1}{2}nonb\_cut)$

worst case scenarios





## Conclusions

## Current Userbase

- Dr Pandit's group at USF
  - silica-water systems
- Dr Strachan's group at Purdue
  - *Si/Ge* nanobar
- Dr Buehler's group at MIT
  - Silica cracking with strain
- Dr van Duin at PennState
- Dr Goddard's group at Caltech
- Dr Aluru's group at UIUC